

What is Titration?

Titration is a quantitative chemical analysis method which involves the slow addition of a solution of a known concentration (titrant) to a known volume of another solution of unknown concentration until the reaction reaches endpoint which is often indicated by a colour change.

Volumetric titration is based on a complete chemical reaction between the analyte and a reagent (titrant) of known concentration which is added to the sample. In a broad sense, titration is a technique to determine the concentration of an unknown solution and has been included in most of the key international pharmacopoeias and other regulations. Titration results are influenced by the following factors:

- Measuring method
- Instrument (instrument error/abrasion of the burette)
- Equipment (electrode error/alteration of electrodes)
- Solution preparation (e.g. preparation of dilution)
- Sample Weighing (weighing error)
- Temperature
- Change of the volumetric solution because of influences through environmental oxygen, carbon dioxide, microorganisms

What are Titration Reactions?

| Type | Theory | Scope | Application |
|----------------|---|---|--------------------------|
| Acid-Base | Proton Transfer $\text{Acid} + \text{Base} \rightleftharpoons \text{Conjugate Base} + \text{Conjugate Acid}$ | Acidimetry ✓ Hydrochloric acid ✓ Sulfuric Acid ✓ Perchloric Acid Alkalimetry ✓ Sodium Hydroxide ✓ Potassium Hydroxide | Saponification value |
| REDOX | Electron Transfer $A_{\text{red}} + B_{\text{ox}} \rightleftharpoons A_{\text{ox}} + B_{\text{red}}$ | Bromatometry ✓ Potassium Bromate Cerimetry ✓ Cerium(IV) sulfate Dichrometry ✓ Chromium (II) sulfate Iodometry ✓ Potassium Triiodate ✓ Sodium Thiosulfate Permanganimetry ✓ Potassium Permanganate Ferrometry ✓ Iron(II) Sulfate | Vitamin C content |
| Precipitation | Formation of sparingly soluble precipitate | Argentometry ✓ Silver Nitrate | Salt as chloride content |
| Complexometric | Metals ions combine with ligands to form soluble and stable complexes | EDTA ✓ Disodium EDTA | Calcium content |

Can Titration be carried out in Different Media?

| Media | Description | Application |
|----------------------|--|--|
| Aqueous | Solvent and Titrant are in aqueous solution | Wide variety of analytes that that soluble in water. |
| Non-Aqueous | Solvent and titrant contain practically no water Aprotic Solvents Benzene, Toluene, Carbon tetrachloride Protophilic Solvents Amine, Ether, Liquid ammonia Protogenic Solvents Formic Acid Amphiprotic Solvents Alcohols Acetic acid | Pharamacopoeial assays |
| Two-Phase Transition | A suitable non-miscible solvent like e.g. chloroform toluene or dichloromethane is added to the aqueous sample solution | Anionic surfactant |

Takween Titrant Quality

TAKWEEN understands the importance of the exact concentration of the titrant. TAKWEEN titrant solution are prepared using ASTM Type 1 water to ensure quality of titrant solutions and highly purity solute.

Takween Titrant Traceability

| No. | Code | Name | Traceability | Provider |
|-----|------|---|--------------|---|
| 1 | 84L | Potassium Hydrogen Phthalate (Acidimetric Primary Standard) | NIST SRM | National Institute of Standards & Technology (NIST) |
| 2 | 136f | Potassium Dichromate (Oxidimetric Standard) | NIST SRM | National Institute of Standards & Technology (NIST) |
| 3 | 185i | Potassium Hydrogen Phthalate (pH Standard) | NIST SRM | National Institute of Standards & Technology (NIST) |
| 4 | 186g | Potassium Dihydrogen Phosphate (186-I-g) Disodium Hydrogen Phosphate (186-II-g) (pH Standard) | NIST SRM | National Institute of Standards & Technology (NIST) |
| 5 | 187e | Sodium Tetraborate Decahydrate (Borax) (pH Standard) | NIST SRM | National Institute of Standards & Technology (NIST) |
| 6 | 188 | Potassium Hydrogen Tartrate (pH Standard) | NIST SRM | National Institute of Standards & Technology (NIST) |
| 7 | 189c | Potassium Tetroxalate Dihydrate (pH Standard) | NIST SRM | National Institute of Standards & Technology (NIST) |
| 8 | 191d | Sodium Bicarbonate (191d-I) Sodium Carbonate (191d-II) (pH Standard) | NIST SRM | National Institute of Standards & Technology (NIST) |
| 9 | 351a | Sodium Carbonate (Acidimetric Standard) | NIST SRM | National Institute of Standards & Technology (NIST) |

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| No. | Code | Name | Traceability | Provider |
|-----|----------|--|--|---|
| 10 | PH216.L5 | pH 4.008 at 25°C Potassium Hydrogen Phtalate (Primary pH-buffer Solution) | Certified Reference Material (CRM) Metrological Traceability: NIST 185i SRM | CPA Chem, ISO 17034:2016, ISO/IEC 17025:2017, ISO 9001:2015 |
| 11 | PH217.L5 | pH 6.865 at 25°C Potassium Dihydrogen Phosphate/ Hydrogen Phosphate (Primary pH-buffer Solution) | Certified Reference Material (CRM) Metrological Traceability: NIST 186g SRM | CPA Chem, ISO 17034:2016, ISO/IEC 17025:2017, ISO 9001:2015 |
| 12 | PH220.L5 | H 10.01 at 25°C Sodium Carbonate/Sodium Hydrogen Carbonate (Primary pH-buffer Solution) | Certified Reference Material (CRM) Metrological Traceability: NIST 191d SRM | CPA Chem, ISO 17034:2016, ISO/IEC 17025:2017, ISO 9001:2015 |

PRODUCT CATALOG

| PRODUCT NAME | PRODUCT CODE |
|--|--------------|
| Hyamine Solution 0.005N | HYNV000050 |
| Hyamine Solution 0.008N | HYNV000080 |
| Hydrochloric Acid 0.01N (0.01M) | HANV000100 |
| Hydrochloric Acid 0.02N (0.02M) | HANV000200 |
| Hydrochloric Acid 0.1000N (0.1000M) in IPA | HINV001000 |
| Hydrochloric Acid 0.10N (0.10M) | HANV001000 |
| Hydrochloric Acid 0.10N (0.10M) | HANV001000 |
| Hydrochloric Acid 0.50N (0.50M) | HANV005000 |
| Hydrochloric Acid 1.00N (1.00M) | HANV010000 |
| Hydrochloric Acid 1.00N (1.00M) | HANV010000 |
| Hydrochloric Acid 10%(w/v) | HAPV01000/WW |
| Hydrochloric Acid 10%wt. | HAPV01000/WW |
| Hydrochloric Acid 10,000ppm | HAAR1000000 |
| Hydrochloric Acid 2.00N (2.00M) | HANV020000 |
| Hydrochloric Acid 25%vol. | HAPV02500/VV |
| Hydrochloric Acid 5%vol. | HAPV00500/VV |
| Hydrochloric Acid 6.00N (6.00M) | HANV060000 |
| Hydrogen Peroxide 0.10%wt. in Methanol | HMPL00010/WW |
| Hydrogen Peroxide 0.70%wt. in Methanol | HMPL00070/WW |
| Hydrogen Peroxide 3%(w/v) | HRPV00300/WW |
| Hydrogen Sulfide 30ppm | HSAR0003000 |
| Hydrogen Sulfide 5ppm | HSAR0000500 |

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| PRODUCT NAME | PRODUCT CODE |
|--|----------------|
| o-Phosphoric Acid 0.02M | PAMV000200 |
| o-Phosphoric Acid 0.10M | PAMV001000 |
| o-Phosphoric Acid 0.20M | PAMV002000 |
| o-Phosphoric Acid 0.30M | PAMV003000 |
| o-Phosphoric Acid 0.60M | PAMV006000 |
| o-Phosphoric Acid 0.69M | PAMV006900 |
| o-Phosphoric Acid 1.00M | PAMV010000 |
| o-Phosphoric Acid 1.50M | PAMV015000 |
| o-Phosphoric Acid 1.55M | PAMV015500 |
| o-Phosphoric Acid 10%(w/v) | PAPV01000/WV |
| o-Phosphoric Acid 10%vol | PAPV01000/VV |
| o-Phosphoric Acid 6.00M | PAMV060000 |
| Perchloric acid 0.10N (0.10M) in Glacial Acetic Acid | PENV001000 |
| Perchloric acid 0.10N (0.10M) in Glacial Acetic Acid | PENV001000 |
| Perchloric acid 0.10N (0.10M) in Glacial Acetic Acid as per USP 39 | PENV001000/USP |
| Perchloric acid 0.10N (0.10M) in IPA | PLNV001000 |
| Perchloric acid 0.50N (0.50M) in Glacial Acetic Acid | PENV005000 |
| Potassium Hydroxide 0.05N (0.05M) in Ethanol | PTNV000500 |
| Potassium Hydroxide 0.05N (0.05M) in Methanol | PMNV000500 |
| Potassium Hydroxide 0.10N (0.10M) | PXNV001000 |
| Potassium Hydroxide 0.10N (0.10M) | PXNV001000 |
| Potassium Hydroxide 0.10N (0.10M) in Ethanol | PTNV001000 |
| Potassium Hydroxide 0.10N (0.10M) in IPA | PINV001000 |
| Potassium Hydroxide 0.10N (0.10M) in Methanol | PMNV001000 |
| Potassium Hydroxide 0.10N (0.10M) in Methanol | PMNV001000 |
| Potassium Hydroxide 0.50N (0.50M) in Ethanol | PTNV005000 |
| Potassium Hydroxide 0.77N (0.77M) | PXNV007700 |
| Potassium Hydroxide 1.00N (1.00M) | PXNV010000 |
| Potassium Hydroxide 1.00N (1.00M) in Methanol | PMNV010000 |
| Potassium Hydroxide 30%(w/v) | PXPV03000/WV |
| Sodium Hydroxide 0.01N (0.01M) | SHNV000100 |
| Sodium Hydroxide 0.05N (0.05M) | SHNV000500 |
| Sodium Hydroxide 0.06N (0.06M) | SHNV000600 |
| Sodium Hydroxide 0.10N (0.10M) | SHNV001000 |
| Sodium Hydroxide 0.10N (0.10M) | SHNV001000 |
| Sodium Hydroxide 0.50N (0.50M) | SHNV005000 |

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| PRODUCT NAME | PRODUCT CODE |
|---|----------------|
| Sodium Hydroxide 0.50N (0.50M) in Ethanol | SXNV005000 |
| Sodium Hydroxide 1.00N (1.00M) | SHNV010000 |
| Sodium Hydroxide 1.00N (1.00M) | SHNV010000 |
| Sodium Hydroxide 1.20N (1.20M) | SHNV012000 |
| Sodium Hydroxide 1.20N (Biotector) | SHNV012000/BIO |
| Sodium Hydroxide 10% (w-v) | SHPV01000/WV |
| Sodium Hydroxide 10.00N (10.00M) | SHNV100000 |
| Sodium Hydroxide 16% (w-v) | SHPV01600/WV |
| Sodium Hydroxide 18%wt. | SHPV01800/WV |
| Sodium Hydroxide 2.00N (2.00M) | SHNV020000 |
| Sodium Hydroxide 2.00N (2.00M) | SHNV020000 |
| Sodium Hydroxide 25,000ppm | SHAR2500000 |
| Sodium Hydroxide 32%wt. | SHPV03200/WV |
| Sodium Hydroxide 4% (w-v) | SHPV00400/WV |
| Sodium Hydroxide 4.00N (4.00M) | SHNV040000 |
| Sodium Hydroxide 4.00N (Biotector) | SHNV040000/BIO |
| Sodium Hydroxide 5% (w-v) | SHPV00500/WV |
| Sodium Hydroxide 50% (w-v) | SHPV05000/WV |
| Sodium Hydroxide 6.00N (6.00M) | SHNV060000 |
| Sodium Hydroxide 6.00N (Biotector) | SHNV060000/BIO |
| Sodium Hydroxide 8.00N (8.00M) | SHNV080000 |
| Sodium Hydroxide Standard for 40mL of 0.10N HCl (8mL x 12vials) | SHNR0050008/12 |
| Sodium Hypochlorite 10.00%wt. available Cl | SYPV01000/WV |
| Sodium Hypochlorite 10ppm available Cl | SYAR0001000 |
| Sodium Hypochlorite 4.00%wt. available Cl | SYPV00400/WV |
| Sodium Hypochlorite 6.36%(w/v) available Cl | SYPV00636/WV |
| Sodium Hypochlorite 6.36%wt. available Cl | SYPV00636/WV |
| Sodium Hypochlorite 80ppm available Cl | SYAR0008000 |
| Sodium Lauryl Sulfate 0.003N | SLNV000030 |
| Sodium Sulfite 0.002N | SFNV000020 |
| Sodium Sulfite 0.02N | SFNV000200 |
| Sodium Sulfite 5%(w/v) | SFPV00500/WV |
| Sodium Sulfite Analyzer 1.0M | SFMV010000 |
| Sodium Thiosulfate 0.002N (0.002M) | STNV000020 |

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| PRODUCT NAME | PRODUCT CODE |
|---|--------------|
| Sodium Thiosulfate 0.005N (0.005M) | STNV000050 |
| Sodium Thiosulfate 0.01N (0.01M) | STNV000100 |
| Sodium Thiosulfate 0.01N (0.01M) | STNV000100 |
| Sodium Thiosulfate 0.0394N (0.0394M) | STNV000394 |
| Sodium Thiosulfate 0.0473N (0.0473M) | STNV000473 |
| Sodium Thiosulfate 0.05N (0.05M) | STNV000500 |
| Sodium Thiosulfate 0.10N (0.10M) | STNV001000 |
| Sodium Thiosulfate 1.00N (1.00M) | STNV010000 |
| Silver Nitrate 0.0141N (0.0141M) | SNNT000141 |
| Silver Nitrate 0.01N (0.01M) | SNNT000100 |
| Silver Nitrate 0.01N (0.01M) | SNNT000100 |
| Silver Nitrate 0.02N (0.02M) | SNNT000200 |
| Silver Nitrate 0.02N (0.02M) | SNNT000200 |
| Silver Nitrate 0.05N (0.05M) | SNNT000500 |
| Silver Nitrate 0.10N (0.10M) | SNNT001000 |
| Silver Nitrate 0.10N (0.10M) | SNNT001000 |
| Silver Nitrate 0.10N (0.10M) in Ethanol | SENV001000 |
| Silver Nitrate 0.171N (0.171M) | SNNT001710 |
| Silver Nitrate 0.282N (0.282M) | SNNT002820 |
| Silver Nitrate 1.00N (1.00M) | SNNT010000 |
| Iodine Solution 0.0473N | IONV000473 |
| Iodine Solution 0.10N | IONV001000 |
| Iodine Solution 1.00N | IONV010000 |
| Iodine Standard Solution 10ppm | IOAR0001000 |